

## Semisynthetic roxburghin tetramethyl ether

Alex Saez,<sup>a,b\*</sup> Carmen Ramirez de Arellano,<sup>c</sup> Noureddine El Aouad,<sup>d</sup> Silvia Rodriguez,<sup>a</sup> Felipe Otalvaro,<sup>a</sup> Diego Cortes<sup>d</sup> and Jairo Saez<sup>a</sup>

<sup>a</sup>Instituto de Química, Química de Plantas Colombianas, Universidad de Antioquia, AA 1226, Medellín, Colombia, <sup>b</sup>Grupo de Procesos Ambientales y Biotecnológicos, Departamento de Ingeniería de Procesos, Universidad EAFIT, AA 3300, Medellín, Colombia, <sup>c</sup>Departamento de Química Orgánica, Universidad de Valencia, E-46100 Valencia, Spain, and <sup>d</sup>Departamento de Farmacología, Facultad de Farmacia, Universidad de Valencia, Burjassot, Valencia, Spain  
Correspondence e-mail: asaez@eafit.edu.co

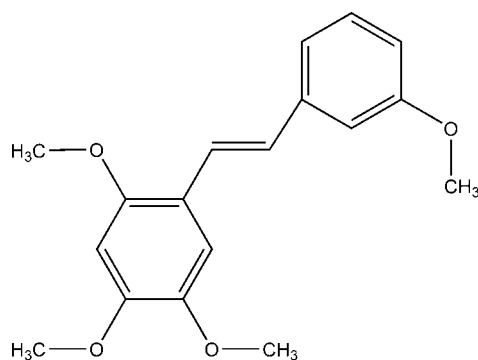
Received 25 February 2008; accepted 16 June 2008

Key indicators: single-crystal X-ray study;  $T = 150$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.048;  $wR$  factor = 0.137; data-to-parameter ratio = 21.6.

The title molecule, (*E*)-2,3',4,5-tetramethoxystilbene,  $\text{C}_{18}\text{H}_{20}\text{O}_4$ , is virtually planar. The angle between the two benzene rings is  $4.06(6)^\circ$ . The intermolecular interactions present in the structure are weak. There are  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds and  $\text{C}-\text{H}\cdots\pi$ -electron ring interactions. The molecules are ordered into planes that are parallel to  $(\bar{1}01)$ . The distance between adjacent planes is about  $3.3$  Å and therefore  $\pi-\pi$  electron interactions between the aromatic planes are also plausible.

## Related literature

For the importance and useful applications of stilbenoid compounds, see: Cushman *et al.* (1991); Nakamura *et al.* (2006). For the precursors of the title compound, see: Krishnamurthy & Maheshwari (1988); Anjaneyulu *et al.* (1990); Wang *et al.* (1988); Murillo (2001).



## Experimental

## Crystal data

$\text{C}_{18}\text{H}_{20}\text{O}_4$   
 $M_r = 300.34$   
 Triclinic,  $P\bar{1}$   
 $a = 7.9633(4)$  Å  
 $b = 9.2454(5)$  Å  
 $c = 11.6194(5)$  Å  
 $\alpha = 73.400(2)^\circ$   
 $\beta = 75.479(3)^\circ$   
 $\gamma = 70.335(2)^\circ$   
 $V = 760.59(7)$  Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 150(2)$  K  
 $0.35 \times 0.10 \times 0.04$  mm

## Data collection

Nonius KappaCCD diffractometer  
 Absorption correction: none  
 8260 measured reflections  
 4391 independent reflections  
 2785 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.136$   
 $S = 0.98$   
 4391 reflections  
 203 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.23$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$  | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{C}7-\text{H}7\cdots\text{O}3$                        | 0.95         | 2.39               | 2.7504 (14) | 102                  |
| $\text{C}17-\text{H}17\text{B}\cdots\text{O}3^{\text{i}}$   | 0.98         | 2.52               | 3.4046 (16) | 150                  |
| $\text{C}18-\text{H}18\text{B}\cdots\text{O}4^{\text{ii}}$  | 0.98         | 2.46               | 3.4342 (15) | 172                  |
| $\text{C}19-\text{H}19\text{B}\cdots\text{O}1^{\text{iii}}$ | 0.98         | 2.51               | 3.4082 (15) | 152                  |
| $\text{C}17-\text{H}17\text{A}\cdots\text{C}g2^{\text{iv}}$ | 0.98         | 2.91               | 3.7863 (15) | 149                  |
| $\text{C}18-\text{H}18\text{C}\cdots\text{C}g1^{\text{v}}$  | 0.98         | 2.67               | 3.5578 (14) | 151                  |

Symmetry codes: (i)  $x, y-1, z$ ; (ii)  $x-1, y-1, z+1$ ; (iii)  $x, y+1, z$ ; (iv)  $-x+1, -y+1, -z$ ; (v)  $-x, -y+1, -z+1$ . Cg1 is the centroid of the C1-C6 ring and Cg2 is the centroid of the C11-C16 ring.

Data collection: COLLECT (Nonius, 1998); cell refinement: SCALEPACK (Otwinowski & Minor, 1997); data reduction: SCALEPACK; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL/PC (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

The authors thank Colciencias and Universidad de Antioquia (Programa de Sostenibilidad) for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FB2091).

## References

- Anjaneyulu, A. S. R., Rani, G. S., Mallavadhani, U. V. & Murthy, Y. L. N. (1990). *Indian J. Chem. Sect. B*, **29**, 219–223.  
 Cushman, M., Nagarathnam, D., Gopal, D., Chakraborti, A. K., Lin, C. M. & Hamel, E. (1991). *J. Med. Chem.* **34**, 2579–2588.  
 Krishnamurthy, H. G. & Maheshwari, N. (1988). *Indian J. Chem. Sect. B*, **27**, 1035–1036.  
 Murillo, J. (2001). *Biota Colomb.* **2**, 49–58.  
 Nakamura, H., Kuroda, H., Saito, H., Suzuki, R., Yamori, T., Maruyama, K. & Haga, T. (2006). *ChemMedChem*, **1**, 729–740.  
 Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.  
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Wang, Z. W., Ma, W. W., McLaughlin, J. L. & Gupta, M. P. (1988). *J. Nat. Prod.* **51**, 382–384.

**supplementary materials**

*Acta Cryst.* (2008). E64, o1305 [ doi:10.1107/S1600536808018266 ]

## Semisynthetic roxburghin tetramethyl ether

A. Saez, C. Ramirez de Arellano, N. El Aouad, S. Rodriguez, F. Otalvaro, D. Cortes and J. Saez

### Comment

Stilbenoid compounds display significant biological activities (Cushman *et al.*, 1991; Nakamura *et al.*, 2006). Resveratrol and its derivatives deserve considerable attention for their physiological properties and their role in defense mechanisms of the higher plants. The Roxburghin tetramethyl ether (*E*)-2,3',4,5,-tetramethoxystilbene) that is an analogue of resveratrol, has been originally obtained by modifications of roxburghin (Krishnamurty & Maheshwari, 1988). It has been completely synthesized by the Perkins modified reaction (Anjaneyulu *et al.*, 1990). In addition to the crystal structure determination, we report an efficient synthesis of this product by the cross-metathesis of 3-methoxystyrene and 2,4,5-trimethoxystyrene, the latter having been obtained as a natural product from the bark of *Duguetia colombiana* (Annonaceae) (Wang *et al.*, 1988; Murillo, 2001).

### Experimental

The catalyst (Grubbs second generation, 9 mg, 0.01 mmol), 2,4,5-trimethoxystyrene (39 mg, 0.2 mmol) and 2-methoxystyrene (277 mg, 2.0 mmol) were dissolved in dry toluene (10 ml). The solution was refluxed under nitrogen for 24 h at 393 K. The compound was purified by a flash column chromatography with silica gel using hexane/ethylacetate 9:1 as an eluent. The title compound (30.0 mg) was obtained as a yellow powder in a yield of 50.0%.

Suitable crystals (pale yellow needles, 0.35 x 0.10 x 0.04 mm average size) were obtained by slow evaporation in a two solvent system (hexane/ethylacetate 1:1). The identity and purity of the obtained compound was confirmed by spectroscopic methods.

(*E*)-1,2,4-trimethoxy-5-(3-methoxystyryl)benzene(Roxburghin tetramethyl ether): pale yellow needles, <sup>1</sup>H-NMR: (CDCl<sub>3</sub>, 300.13 MHz, numeration according to ellipsoid plot) δ 7.42 (d, *J*= 16.4 Hz, H-7), 7.26 (dd, *J* = 8.3, 7.7 Hz, H-15), 7.12 (s, H-6), 7.12 (d, *J*= 7.7 Hz, H-16), 7.06 (s, H-12), 6.79 (d, *J*= 8.3 Hz, H-14), 6.54 (s, H-3), 3.92 (s, C-2-OCH<sub>3</sub>), 3.92 (s, C-1-OCH<sub>3</sub>), 3.87 (s, C-4-OCH<sub>3</sub>), 3.85 (s, C-13-OCH<sub>3</sub>); <sup>13</sup>C (CDCl<sub>3</sub>, 75.47 MHz) δ 160.2 (C-13), 152.2 (C-4), 150.1 (C-2), 143.8 (C-1), 140.0 (C-11), 129.9 (C-15), 127.1 (C-8), 123.7 (C-7), 119.5 (C-16), 118.6 (C-5), 113.1 (C-12), 111.9 (C-6), 109.8 (C-14), 98.1 (C-3), 57.1 (C-17), 56.9 (C-18), 56.5 (C-19), 55.7 (C-20). EIMS *m/z* 300 (100), 257 (8), 195 (12).

### Refinement

All the H atoms were discernible in the difference electron-density maps. However, they were situated into idealized positions and constrained by riding model approximation. C—H<sub>methyl</sub>=0.98 Å; C—H<sub>aryl</sub>=0.95 Å; *U*<sub>iso</sub>H<sub>methyl</sub>=1.5*U*<sub>eq</sub>(C<sub>methyl</sub>); *U*<sub>iso</sub>H<sub>aryl</sub>=1.2*U*<sub>eq</sub>(C<sub>aryl</sub>).

## Figures

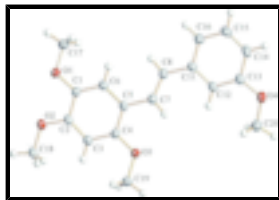


Fig. 1. The title molecule with the displacement ellipsoids shown at the 50% probability level.

## (*E*)-2,3',4,5-tetramethoxystilbene

### Crystal data

$C_{18}H_{20}O_4$

$M_r = 300.34$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 7.9633$  (4) Å

$b = 9.2454$  (5) Å

$c = 11.6194$  (5) Å

$\alpha = 73.400$  (2)°

$\beta = 75.479$  (3)°

$\gamma = 70.335$  (2)°

$V = 760.59$  (7) Å<sup>3</sup>

$Z = 2$

$F_{000} = 320$

$D_x = 1.311$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 46078 reflections

$\theta = 1.0$ – $30.0$ °

$\mu = 0.09$  mm<sup>-1</sup>

$T = 150$  (2) K

Needle, yellow

$0.35 \times 0.10 \times 0.04$  mm

### Data collection

Nonius KappaCCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 9 pixels mm<sup>-1</sup>

$T = 150$ (2) K

$\omega$  scans

Absorption correction: none

8260 measured reflections

4391 independent reflections

2785 reflections with  $I > 2\sigma(I)$

$R_{int} = 0.025$

$\theta_{max} = 30.0$ °

$\theta_{min} = 2.7$ °

$h = -11 \rightarrow 11$

$k = -13 \rightarrow 12$

$l = -16 \rightarrow 16$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.047$

$wR(F^2) = 0.136$

$S = 0.98$

Secondary atom site location: difference Fourier map

Hydrogen site location: difference Fourier map

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0789P)^2]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} < 0.001$

4391 reflections  $\Delta\rho_{\max} = 0.29 \text{ e } \text{\AA}^{-3}$   
 203 parameters  $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$   
 76 constraints Extinction correction: none  
 Primary atom site location: structure-invariant direct methods

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|      | <i>x</i>     | <i>y</i>     | <i>z</i>      | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|---------------|----------------------------------|
| O1   | 0.35419 (11) | 0.13391 (9)  | 0.41479 (8)   | 0.0342 (2)                       |
| O2   | 0.16009 (11) | 0.25274 (9)  | 0.59974 (7)   | 0.0302 (2)                       |
| O3   | 0.31164 (11) | 0.74440 (9)  | 0.41175 (7)   | 0.0322 (2)                       |
| O4   | 0.83171 (12) | 1.06354 (9)  | -0.08834 (8)  | 0.0352 (2)                       |
| C1   | 0.34968 (15) | 0.28542 (12) | 0.40914 (10)  | 0.0257 (2)                       |
| C2   | 0.24495 (14) | 0.34919 (12) | 0.50941 (9)   | 0.0242 (2)                       |
| C3   | 0.23241 (14) | 0.50121 (12) | 0.51169 (10)  | 0.0252 (2)                       |
| H3   | 0.1620       | 0.5441       | 0.5796        | 0.030*                           |
| C4   | 0.32276 (14) | 0.59219 (12) | 0.41457 (10)  | 0.0242 (2)                       |
| C5   | 0.42856 (14) | 0.53130 (12) | 0.31388 (10)  | 0.0237 (2)                       |
| C6   | 0.43875 (15) | 0.37648 (12) | 0.31392 (10)  | 0.0260 (2)                       |
| H6   | 0.5092       | 0.3330       | 0.2463        | 0.031*                           |
| C7   | 0.52275 (15) | 0.62808 (13) | 0.21313 (10)  | 0.0255 (2)                       |
| H7   | 0.5008       | 0.7340       | 0.2176        | 0.031*                           |
| C8   | 0.63618 (15) | 0.58230 (13) | 0.11577 (10)  | 0.0286 (2)                       |
| H8   | 0.6583       | 0.4762       | 0.1116        | 0.034*                           |
| C11  | 0.73066 (15) | 0.67918 (13) | 0.01411 (10)  | 0.0266 (2)                       |
| C12  | 0.72987 (14) | 0.83038 (13) | 0.01567 (10)  | 0.0257 (2)                       |
| H12  | 0.6661       | 0.8741       | 0.0844        | 0.031*                           |
| C13  | 0.82170 (15) | 0.91678 (13) | -0.08266 (10) | 0.0275 (3)                       |
| C14  | 0.91383 (17) | 0.85479 (15) | -0.18482 (11) | 0.0352 (3)                       |
| H14  | 0.9754       | 0.9144       | -0.2524       | 0.042*                           |
| C15  | 0.91459 (18) | 0.70674 (15) | -0.18674 (11) | 0.0400 (3)                       |
| H15  | 0.9770       | 0.6642       | -0.2562       | 0.048*                           |
| C16  | 0.82497 (17) | 0.61818 (14) | -0.08813 (11) | 0.0348 (3)                       |
| H16  | 0.8281       | 0.5154       | -0.0906       | 0.042*                           |
| C17  | 0.48206 (17) | 0.05869 (13) | 0.32371 (12)  | 0.0372 (3)                       |
| H17A | 0.4483       | 0.1127       | 0.2435        | 0.056*                           |

## supplementary materials

---

|      |              |              |              |            |
|------|--------------|--------------|--------------|------------|
| H17B | 0.4825       | -0.0516      | 0.3411       | 0.056*     |
| H17C | 0.6030       | 0.0634       | 0.3241       | 0.056*     |
| C18  | 0.06092 (16) | 0.31142 (13) | 0.70597 (10) | 0.0303 (3) |
| H18A | 0.1428       | 0.3348       | 0.7439       | 0.046*     |
| H18B | 0.0074       | 0.2319       | 0.7640       | 0.046*     |
| H18C | -0.0353      | 0.4077       | 0.6828       | 0.046*     |
| C19  | 0.19201 (16) | 0.81424 (13) | 0.50742 (11) | 0.0329 (3) |
| H19A | 0.0689       | 0.8124       | 0.5097       | 0.049*     |
| H19B | 0.1939       | 0.9235       | 0.4930       | 0.049*     |
| H19C | 0.2309       | 0.7550       | 0.5854       | 0.049*     |
| C20  | 0.75180 (17) | 1.12894 (14) | 0.01691 (11) | 0.0349 (3) |
| H20A | 0.6208       | 1.1449       | 0.0326       | 0.052*     |
| H20B | 0.7762       | 1.2302       | 0.0029       | 0.052*     |
| H20C | 0.8036       | 1.0566       | 0.0875       | 0.052*     |

### Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$   | $U^{22}$   | $U^{33}$   | $U^{12}$    | $U^{13}$    | $U^{23}$    |
|-----|------------|------------|------------|-------------|-------------|-------------|
| O1  | 0.0446 (5) | 0.0235 (4) | 0.0337 (5) | -0.0156 (3) | 0.0077 (4)  | -0.0104 (3) |
| O2  | 0.0344 (4) | 0.0262 (4) | 0.0267 (4) | -0.0133 (3) | 0.0047 (3)  | -0.0040 (3) |
| O3  | 0.0395 (5) | 0.0247 (4) | 0.0317 (4) | -0.0148 (3) | 0.0080 (4)  | -0.0108 (3) |
| O4  | 0.0432 (5) | 0.0296 (4) | 0.0328 (5) | -0.0187 (4) | 0.0041 (4)  | -0.0064 (4) |
| C1  | 0.0290 (6) | 0.0202 (5) | 0.0279 (6) | -0.0089 (4) | -0.0023 (5) | -0.0054 (4) |
| C2  | 0.0240 (5) | 0.0245 (5) | 0.0221 (5) | -0.0092 (4) | -0.0019 (4) | -0.0015 (4) |
| C3  | 0.0251 (5) | 0.0255 (5) | 0.0243 (5) | -0.0078 (4) | -0.0008 (4) | -0.0066 (4) |
| C4  | 0.0256 (5) | 0.0212 (5) | 0.0264 (6) | -0.0083 (4) | -0.0024 (4) | -0.0062 (4) |
| C5  | 0.0231 (5) | 0.0239 (5) | 0.0242 (5) | -0.0089 (4) | -0.0024 (4) | -0.0040 (4) |
| C6  | 0.0278 (6) | 0.0255 (5) | 0.0244 (5) | -0.0095 (4) | 0.0002 (4)  | -0.0069 (4) |
| C7  | 0.0272 (6) | 0.0238 (5) | 0.0255 (6) | -0.0097 (4) | -0.0024 (5) | -0.0045 (4) |
| C8  | 0.0339 (6) | 0.0233 (5) | 0.0279 (6) | -0.0116 (4) | 0.0006 (5)  | -0.0056 (4) |
| C11 | 0.0261 (5) | 0.0276 (6) | 0.0244 (5) | -0.0091 (4) | -0.0007 (4) | -0.0046 (4) |
| C12 | 0.0259 (5) | 0.0282 (5) | 0.0217 (5) | -0.0089 (4) | 0.0002 (4)  | -0.0059 (4) |
| C13 | 0.0274 (6) | 0.0281 (6) | 0.0266 (6) | -0.0105 (5) | -0.0020 (5) | -0.0048 (5) |
| C14 | 0.0397 (7) | 0.0398 (7) | 0.0259 (6) | -0.0210 (6) | 0.0055 (5)  | -0.0051 (5) |
| C15 | 0.0488 (8) | 0.0436 (7) | 0.0278 (6) | -0.0192 (6) | 0.0094 (6)  | -0.0149 (6) |
| C16 | 0.0434 (7) | 0.0302 (6) | 0.0309 (6) | -0.0158 (5) | 0.0058 (5)  | -0.0112 (5) |
| C17 | 0.0439 (7) | 0.0279 (6) | 0.0381 (7) | -0.0126 (5) | 0.0071 (6)  | -0.0141 (5) |
| C18 | 0.0343 (6) | 0.0345 (6) | 0.0211 (5) | -0.0152 (5) | 0.0019 (5)  | -0.0038 (5) |
| C19 | 0.0353 (6) | 0.0276 (6) | 0.0358 (7) | -0.0111 (5) | 0.0055 (5)  | -0.0142 (5) |
| C20 | 0.0374 (7) | 0.0310 (6) | 0.0373 (7) | -0.0134 (5) | 0.0003 (5)  | -0.0104 (5) |

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

|        |             |         |             |
|--------|-------------|---------|-------------|
| O1—C1  | 1.3720 (12) | C11—C12 | 1.4011 (15) |
| O1—C17 | 1.4298 (13) | C12—C13 | 1.3896 (15) |
| O2—C2  | 1.3670 (13) | C12—H12 | 0.9500      |
| O2—C18 | 1.4308 (13) | C13—C14 | 1.3945 (16) |
| O3—C4  | 1.3713 (12) | C14—C15 | 1.3733 (16) |
| O3—C19 | 1.4231 (13) | C14—H14 | 0.9500      |

|             |             |               |             |
|-------------|-------------|---------------|-------------|
| O4—C13      | 1.3677 (13) | C15—C16       | 1.3927 (17) |
| O4—C20      | 1.4281 (13) | C15—H15       | 0.9500      |
| C1—C6       | 1.3828 (15) | C16—H16       | 0.9500      |
| C1—C2       | 1.4052 (14) | C17—H17A      | 0.9800      |
| C2—C3       | 1.3826 (14) | C17—H17B      | 0.9800      |
| C3—C4       | 1.3988 (15) | C17—H17C      | 0.9800      |
| C3—H3       | 0.9500      | C18—H18A      | 0.9800      |
| C4—C5       | 1.3994 (14) | C18—H18B      | 0.9800      |
| C5—C6       | 1.4060 (14) | C18—H18C      | 0.9800      |
| C5—C7       | 1.4651 (15) | C19—H19A      | 0.9800      |
| C6—H6       | 0.9500      | C19—H19B      | 0.9800      |
| C7—C8       | 1.3336 (16) | C19—H19C      | 0.9800      |
| C7—H7       | 0.9500      | C20—H20A      | 0.9800      |
| C8—C11      | 1.4721 (15) | C20—H20B      | 0.9800      |
| C8—H8       | 0.9500      | C20—H20C      | 0.9800      |
| C11—C16     | 1.3942 (15) |               |             |
| C1—O1—C17   | 116.35 (8)  | O4—C13—C14    | 115.01 (9)  |
| C2—O2—C18   | 117.08 (8)  | C12—C13—C14   | 120.36 (10) |
| C4—O3—C19   | 117.81 (8)  | C15—C14—C13   | 119.33 (10) |
| C13—O4—C20  | 117.84 (8)  | C15—C14—H14   | 120.3       |
| O1—C1—C6    | 125.08 (10) | C13—C14—H14   | 120.3       |
| O1—C1—C2    | 115.75 (9)  | C14—C15—C16   | 120.85 (11) |
| C6—C1—C2    | 119.18 (9)  | C14—C15—H15   | 119.6       |
| O2—C2—C3    | 124.25 (10) | C16—C15—H15   | 119.6       |
| O2—C2—C1    | 115.93 (9)  | C15—C16—C11   | 120.49 (10) |
| C3—C2—C1    | 119.82 (9)  | C15—C16—H16   | 119.8       |
| C2—C3—C4    | 120.40 (10) | C11—C16—H16   | 119.8       |
| C2—C3—H3    | 119.8       | O1—C17—H17A   | 109.5       |
| C4—C3—H3    | 119.8       | O1—C17—H17B   | 109.5       |
| O3—C4—C3    | 122.49 (9)  | H17A—C17—H17B | 109.5       |
| O3—C4—C5    | 116.58 (9)  | O1—C17—H17C   | 109.5       |
| C3—C4—C5    | 120.93 (9)  | H17A—C17—H17C | 109.5       |
| C4—C5—C6    | 117.45 (9)  | H17B—C17—H17C | 109.5       |
| C4—C5—C7    | 120.20 (9)  | O2—C18—H18A   | 109.5       |
| C6—C5—C7    | 122.35 (10) | O2—C18—H18B   | 109.5       |
| C1—C6—C5    | 122.22 (10) | H18A—C18—H18B | 109.5       |
| C1—C6—H6    | 118.9       | O2—C18—H18C   | 109.5       |
| C5—C6—H6    | 118.9       | H18A—C18—H18C | 109.5       |
| C8—C7—C5    | 126.52 (10) | H18B—C18—H18C | 109.5       |
| C8—C7—H7    | 116.7       | O3—C19—H19A   | 109.5       |
| C5—C7—H7    | 116.7       | O3—C19—H19B   | 109.5       |
| C7—C8—C11   | 126.74 (10) | H19A—C19—H19B | 109.5       |
| C7—C8—H8    | 116.6       | O3—C19—H19C   | 109.5       |
| C11—C8—H8   | 116.6       | H19A—C19—H19C | 109.5       |
| C16—C11—C12 | 118.51 (10) | H19B—C19—H19C | 109.5       |
| C16—C11—C8  | 118.73 (10) | O4—C20—H20A   | 109.5       |
| C12—C11—C8  | 122.76 (10) | O4—C20—H20B   | 109.5       |
| C13—C12—C11 | 120.45 (10) | H20A—C20—H20B | 109.5       |
| C13—C12—H12 | 119.8       | O4—C20—H20C   | 109.5       |

## supplementary materials

---

|             |             |               |       |
|-------------|-------------|---------------|-------|
| C11—C12—H12 | 119.8       | H20A—C20—H20C | 109.5 |
| O4—C13—C12  | 124.63 (10) | H20B—C20—H20C | 109.5 |

### Hydrogen-bond geometry (Å, °)

| <i>D</i> —H $\cdots$ <i>A</i>       | <i>D</i> —H | H $\cdots$ <i>A</i> | <i>D</i> $\cdots$ <i>A</i> | <i>D</i> —H $\cdots$ <i>A</i> |
|-------------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C7—H7 $\cdots$ O3                   | 0.95        | 2.39                | 2.7504 (14)                | 102                           |
| C17—H17B $\cdots$ O3 <sup>i</sup>   | 0.98        | 2.52                | 3.4046 (16)                | 150                           |
| C18—H18B $\cdots$ O4 <sup>ii</sup>  | 0.98        | 2.46                | 3.4342 (15)                | 172                           |
| C19—H19B $\cdots$ O1 <sup>iii</sup> | 0.98        | 2.51                | 3.4082 (15)                | 152                           |
| C17—H17A $\cdots$ Cg2 <sup>iv</sup> | 0.98        | 2.91                | 3.7863 (15)                | 149                           |
| C18—H18C $\cdots$ Cg1 <sup>v</sup>  | 0.98        | 2.67                | 3.5578 (14)                | 151                           |

Symmetry codes: (i)  $x, y-1, z$ ; (ii)  $x-1, y-1, z+1$ ; (iii)  $x, y+1, z$ ; (iv)  $-x+1, -y+1, -z$ ; (v)  $-x, -y+1, -z+1$ .



Fig. 1

